



Standard Test Methods for Polyurethane Raw Materials: Determination of Gardner and APHA Color of Polyols¹

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1. Scope*

1.1 These test methods measure visually the color of clear polyester and polyether liquids. They apply only to materials whose colors have light-absorption characteristics similar to those of the standards. An alternative method is Test Method **D 1209** (see **Note 1**).

1.2 *This standard does not purport to address all of the safety concerns, if any, associated with its use. It is the responsibility of the user of this standard to establish appropriate safety and health practices and determine the applicability of regulatory limitations prior to use.*

NOTE 1—Although Test Method A of these test methods and **ISO 4630-1:2004** differ in some details, data obtained using either are technically equivalent. Although Test Method B of these test methods and **ISO 6271-1981** differ in some details, data obtained using either are technically equivalent.

2. Referenced Documents

2.1 ASTM Standards:²

- D 883** Terminology Relating to Plastics
- D 1193** Specification for Reagent Water
- D 1209** Test Method for Color of Clear Liquids (Platinum-Cobalt Scale)
- D 5386** Test Method for Color of Liquids Using Tristimulus Colorimetry
- E 308** Practice for Computing the Colors of Objects by Using the CIE System
- E 1164** Practice for Obtaining Spectrometric Data for Object-Color Evaluation

2.2 ISO Standards:³

ISO 4630-1:2004 Estimation of Colour by the Gardner Colour Scale—Part 1: Visual Method

ISO 4630-2:2004 Clear Liquids—Estimation of Colour by the Gardner Colour Scale—Part 2: Spectroscopic Method

ISO 6271-1981 Clear Liquids—Estimation of Color by the Platinum-Cobalt Scale

3. Terminology

3.1 For definitions of terms used in these test methods see Terminology **D 883**.

4. Summary of Test Method

4.1 In Test Method A, the color of the material to be tested is compared to a series of color standards with defined chromicity coordinates, prepared on one of three ways. The results are reported as the color standard, which best matches the sample.

4.2 In Test Method B, the color of the material to be tested is compared to a series of platinum-cobalt color standards, designated by mg of Pt/mL of standard solution. The results are reported as the color standard, which best matches the sample (see **Note 2**).

NOTE 2—Color of liquids also can be measured by visible spectroscopy and the results converted to any of several color scales. These results can be converted to the APHA scale by appropriate manipulations, as for example in Test Method **D 5386**.

5. Significance and Use

5.1 These test methods are suitable for quality control, as specification tests, and for research. Color is an important property of urethane products.

6. Sampling

6.1 Polyesters and polyethers usually contain molecules covering an appreciable range of molecular weights. These have a tendency to fractionate during solidification. Unless the material is a liquid or finely ground solid it is necessary to melt (using no higher temperature than necessary) and mix the resin well before removing a sample for analysis. Many polyols are

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² For referenced ASTM standards, visit the ASTM website, www.astm.org, or contact ASTM Customer Service at service@astm.org. For *Annual Book of ASTM Standards* volume information, refer to the standard's Document Summary page on the ASTM website.

³ Available from American National Standards Institute (ANSI), 25 W. 43rd St., 4th Floor, New York, NY 10036.

*A Summary of Changes section appears at the end of this standard.

TABLE 1 Gardner Reference Standard Color Solutions

Gardner Color Standard Number	Chromaticity Coordinates ^A		Potassium Chloroplatinate, g/1000 mL of 0.1 N HCl	Iron-Cobalt Solutions			Potassium Dichromate, g/100 mL Sulfuric Acid ^B
	x	y		Ferric Chloride Solution, mL	Cobalt Chloride Solution, mL	Hydrochloric Acid, mL	
1	0.3190	0.3271	0.550	0.0039
2	0.3241	0.3344	0.865	0.0048
3	0.3315	0.3456	1.330	0.0071
4	0.3433	0.3632	2.080	0.0112
5	0.3578	0.3820	3.035	0.0205
6	0.3750	0.4047	4.225	0.0322
7	0.4022	0.4360	6.400	0.0384
8	0.4179	0.4535	7.900	0.0515
9	0.4338	0.4648	...	3.8	3.0	93.2	0.0780
10	0.4490	0.4775	...	5.1	3.6	91.3	0.164
11	0.4836	0.4805	...	7.5	5.3	87.2	0.250
12	0.5084	0.4639	...	10.8	7.6	81.6	0.380
13	0.5395	0.4451	...	16.6	10.0	73.4	0.572
14	0.5654	0.4295	...	22.2	13.3	64.5	0.763
15	0.5870	0.4112	...	29.4	17.6	53.0	1.041
16	0.6060	0.3933	...	37.8	22.8	39.4	1.280
17	0.6275	0.3725	...	51.3	25.6	23.1	2.220
18	0.6475	0.3525	...	100.0	0.0	0.0	3.00

^A Chromaticity coordinates for CIE standard illuminant C and the CIE 1931 (2°) standard observer.

^B The dichromate color standards have been found to be less reliable than chloroplatinate or iron-cobalt color standards. They are included in **Table 1** for reference only.

hygroscopic and care should be taken to provide minimum exposure to atmospheric moisture during the sampling.

7. Purity of Reagents

7.1 *Purity of Reagents*—Reagent-grade chemicals shall be used in all tests. Unless otherwise indicated, it is intended that all reagents shall conform to the specifications of the Committee on Analytical Reagents of the American Chemical Society, where such specifications are available.⁴ Other grades may be used, provided it is first ascertained that the reagent is of sufficiently high purity to permit its use without lessening the accuracy of the determination.

7.2 *Purity of Water*—Unless otherwise indicated, references to water shall be understood to mean reagent water as defined by Type IV or better of Specification **D 1193**.

TEST METHOD A—GARDNER COLOR

8. Apparatus

8.1 *Gardner-Holdt Tubes*, of clear glass, with closed, flat, even bottoms, and having the following approximate dimensions and markings:

8.1.1 A uniform internal length of 112 mm,

8.1.2 A uniform internal diameter throughout the length of the tube of 10.75 mm, and

8.1.3 An etched line around the outside of the tube 5 mm from the open end and a second etched line around the outside of the tube 13 mm from the open end.

⁴ *Reagent Chemicals, American Chemical Society Specifications*, American Chemical Society, Washington, DC. For suggestions on the testing of reagents not listed by the American Chemical Society, see *Analar Standards for Laboratory Chemicals*, BDH Ltd., Poole, Dorset, U.K., and the *United States Pharmacopeia and National Formulary*, U.S. Pharmacopeial Convention, Inc. (USPC), Rockville, MD.

9. Reagents

9.1 *Cobalt Chloride Solution*—Prepare a solution containing 1 part by weight of cobalt chloride ($\text{CoCl}_2 \times 6\text{H}_2\text{O}$) to 3 parts of HCl (1 to 17).

9.2 *Ferric Chloride Solution*—Prepare a solution containing approximately 5 parts by weight of ferric chloride ($\text{FeCl}_3 \times 6\text{H}_2\text{O}$) and 1.2 parts of HCl (1 to 17). Adjust to exact color equivalence to a freshly prepared solution containing 3 g of $\text{K}_2\text{Cr}_2\text{O}_7$ in 100 mL of H_2SO_4 (sp gr 1.84).

9.3 *Hydrochloric Acid (1 to 17)*—Mix 1 volume of concentrated hydrochloric acid (HCl, sp gr 1.19) with 17 volumes of water.

9.4 *Hydrochloric Acid (0.1 N)*—Prepare 0.1 N HCl.

9.5 *Potassium Chloroplatinate* ($\text{K}_2\text{Cr}_2\text{O}_7$).

9.6 *Potassium Dichromate* ($\text{K}_2\text{Cr}_2\text{O}_7$).

9.7 *Sulfuric Acid* (sp gr 1.84)—Concentrated sulfuric acid (H_2SO_4).

10. Gardner Color Reference Standards⁵

10.1 The primary standards for color shall consist of solutions defined by their spectral transmittance in 1-cm cell with parallel sides. The chromaticity coordinates of these solutions shall conform to those given in **Table 1** when determined on a 1-cm layer of the solution in accordance with Practice **E 1164** and Test Method **E 308**.

10.2 For comparison, permanent solutions of known color are more satisfactory. The approximate composition of solutions giving each of the 18 Gardner colors is also given in **Table 1**. The solutions shall be made from K_2PtCl_6 in 0.1 N

⁵ The sole source of supply of the glass color standards and color standard solutions known to the committee at this time is BYK-Gardner USA, 9104 Guilford Road, Columbia, MD 21046. If you are aware of alternative suppliers, please provide this information to ASTM International Headquarters. Your comments will receive careful consideration at a meeting of the responsible technical committee,¹ which you may attend.